

# **PLASMA TREATED HERBAL WOUND DRESSING**

**A PROJECT WORK**

*Submitted by*

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## BONAFIDE CERTIFICATE

Certified that this project report “**PLASMA TREATED HERBAL WOUND DRESSING**” is the bonafide work of “**S.GOWRI, A.NITHYA PRIYA, D.SIVARAJ, M.MOHAMED ASHIF**” who carried out the project work under my supervision.



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## **ABSTRACT**

In this project, we are fabricating the wound dressing material from Bamboo and cotton fibers and then extract will be prepared from herbal plant (TULSI). The Herbal plant naturally contains antimicrobial properties .To treat the cotton and bamboo with plasma. Then Application of herbal extract on plasma Treated material and untreated material and assess the antimicrobial activity. Advantage of this project is, it easily and quickly cures wound. Finally we check how effectively plasma treated herbal material works.

**INTRODUCTION**



## **CHAPTER 1**

### **INTRODUCTION**

Medical textile is one of the fastest growing sectors of the technical textiles industry.

Many researches and developments are coming day by day. In this project “**HERBAL FINISHING FOR WOUND DRESSINGS**”, herbs which are not known by many people are being used.

The commonly available herbs, having antimicrobial property in nature are taken for this project. They are fabricated with cotton and bamboo and thus the wound dressings are made.

#### **1.1 OBJECTIVES**

- Herbal Extraction is prepared from the following Plant.

##### **OCIMUM SANCTUM**

- To treat the Cotton and bamboo with plasma.
- Application of herbal extract on plasma Treated material and untreated material.
- Assessment of the antimicrobial activity.
- Influence of plasma treatment on antimicrobial property of the knitted cotton and bamboo fabrics.
- Compare the both the material characteristics.

**LITERATURE REVIEW**

## CHAPTER 2

### 2.1 OCIMUM SANCTUM

This leaf extract has wound-healing activity in Sprague dawley rats.

#### Uses

- It is considered as India's Queen of herbs. They are largely used in ayurvedic medicines. It has medicinal properties as well as cosmetic properties.
- Water boiled with tulsi leaves is good for sore throat. It can also be gargled.
- Chewing tulsi leaves treats cold and flu.
- Tulsi leaf when eaten in the morning purifies blood.
- It can be used as tooth powder by drying its leaves and mixed with water.
- It helps in protecting the entire respiratory tract.
- Its has many cosmetic properties and used in herbal soap and also for body scrub. Tulsi oil can be used for controlling dandruff.
- The objective of our study was to evaluate the antimicrobial and wound healing activity of the leaf extract of OCIMUM SANCTUM



Fig 2.1

## 2.2 ABOUT COTTON

- Cotton is a natural fibre and it is harvested from the cotton plant.
- The properties of cotton are many – it is soft, versatile and strong – to mention a few.
- These qualities make it ideal for clothing and many other items.
- In fact, no other material is quite like cotton.



Fig 2.2

### The most important properties are:

- It is soft
- It “breathes”
- It absorbs body moisture
- It is comfortable
- It is strong and durable
- It is versatile
- It performs well
- It has good colour retention
- It is easy to print on
- It is easy to care for, easy to wash
- It is a natural resource that is fully renewable.

### 2.3 ABOUT BAMBOO

Bamboo fiber is praised as the natural green and eco-friendly new type textile material of 21<sup>st</sup> century. Bamboo is called “THE NEW COTTON” and people are talking about “TEXTILE REVOLUTION”. Bamboo fibers are the newest thing to hit textile area and have a number of applications. Bamboo fiber and starch pulp are made from bamboo that grows widely through Asian countries. Starchy pulp is refined product from bamboo stems and leaves through a process of hydrolysis-alkalization and multi phase bleaching.

Because the cross-section of the fiber is filled with various micro-gaps and micro-holes, it has much better moisture absorption and ventilation. Moisture absorbency is twice than that of cotton with extraordinary soil release.

Natural antibacterial elements (bio-agent “bamboo kun”) in bamboo fiber keep bacteria away from bamboo fabrics.

Bamboo grows quickly requiring few farming inputs and no pesticides. When compared to other fibers such as cotton it is more sustainable. Bamboo takes up more green house gases and releases more oxygen and does not need a replanting or fertilizer. [www.bamboofabrics.com](http://www.bamboofabrics.com)

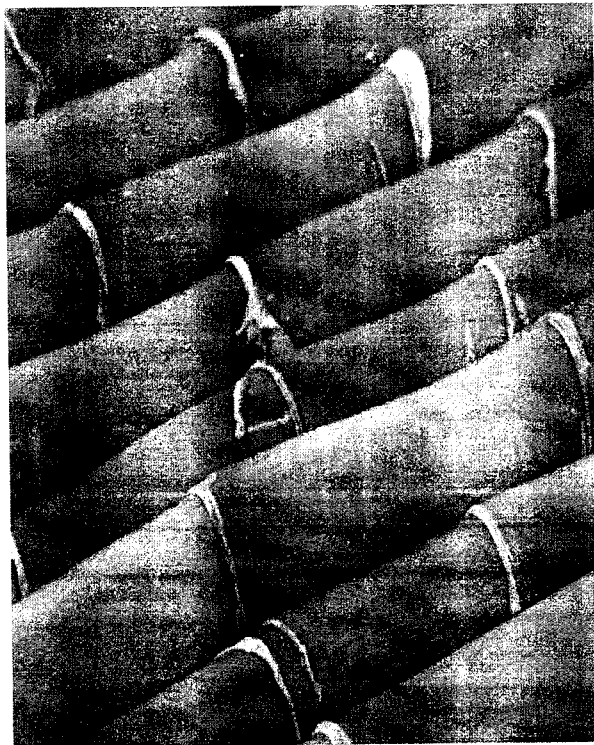


Fig 2.3

## 2.4 WOUND HEALING

Wound is defined simply as the disruption of the cellular and anatomic continuity of a tissue. Wound may be produced by physical, chemical, thermal, microbial or insult to the tissue.

Wound healing is a complex and dynamic process that varies according to the type and location of the wound.

Surgical wounds closed by sutures require very little remodeling whereas larger wounds such immunological as pressure ulcers require considerable tissue reconstruction. Generally however, from injury to resolution, wounds go through four phases...

1. Homeostasis -The stopping of bleeding.
2. Inflammation -Whilst the damaged tissue is preventing further blood loss by vasoconstriction and clotting.
3. Proliferative -Homeostasis and inflammation have contained the emergency and stabilized the situation and now the body needs to start the repair process.
4. Remodeling.

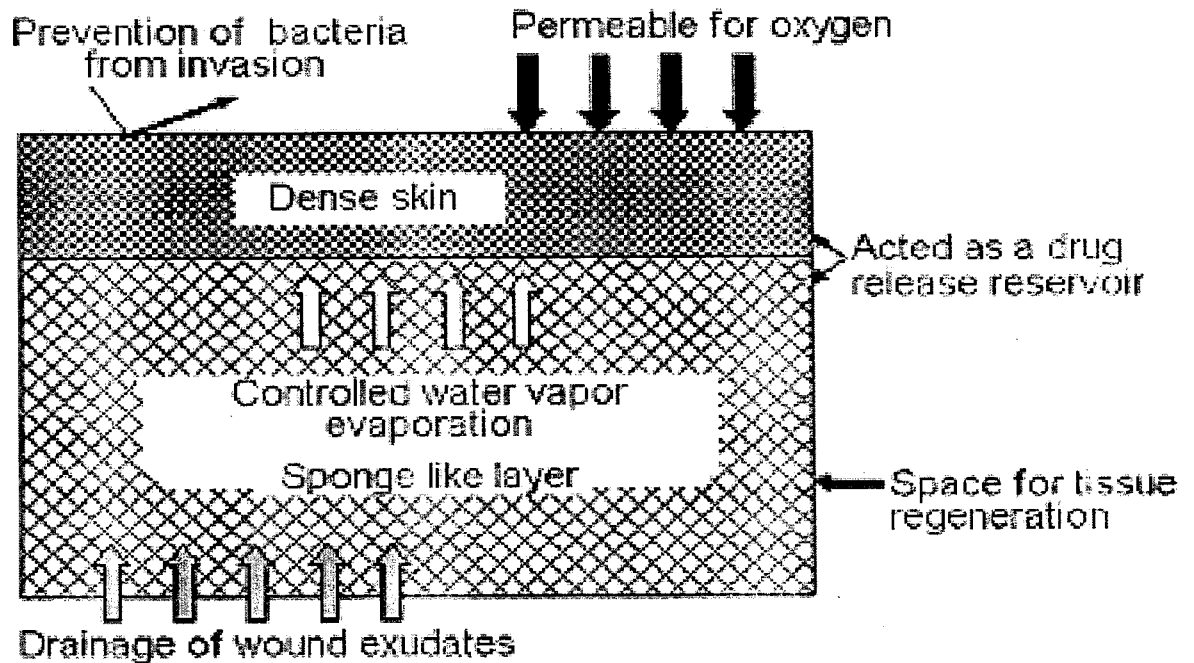


Fig 2.4

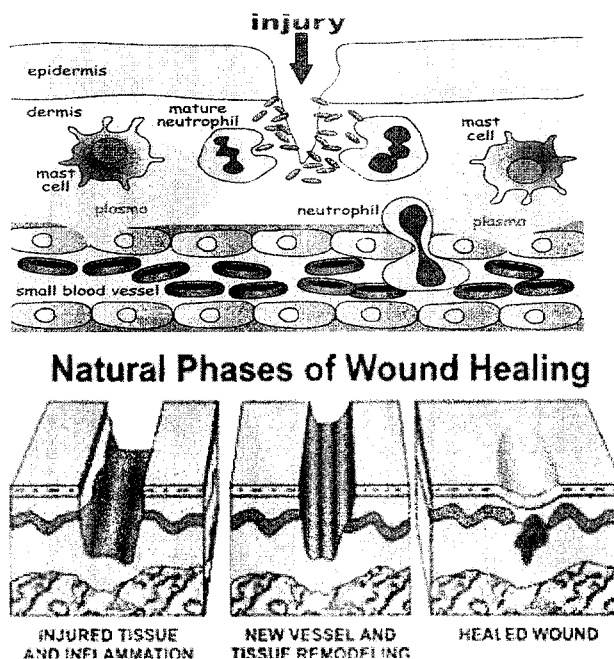


Fig 2.5

### Evaluation of wound healing potential of *Bauhinia purpurea* leaf extracts in rats

The present study was carried out to evaluate the effect of methanol and chloroform extracts *Bauhinia purpurea* on experimentally induced excision, incision, burn and dead space wound models Sprague Dawley rats.

Formulations of methanol and chloroform extracts of *Bauhinia purpurea* were prepared in carbop and simple ointment base at concentrations of 2.5% and 5% and applied to the wounds. In the excision and burn wound models, animals treated with high doses of methanol and chloroform showed significant reduction in time taken for epithelization and wound contraction (50%) compared to control.

A significant increase in breaking strength was found in incision wound model with methanol and chloroform extracts compared to their respective bases. In the dead space wound model, methanol and

chloroform extract treatment (100 and 500 mg/kg) orally produced a significant increase in the breaking strength, dry tissue weight and hydroxyproline content of the granulation tissue when compared to control.

Among the extracts, methanol extract exhibited more activity followed by the chloroform extract. In conclusion, the present study indicated that *Bauhinia purpurea* leaves exhibited wound healing activity.

## 2.5 PLASMA TREATMENT IN TEXTILE

### 2.5.1 Introduction:

- Plasma, the so called fourth state of matter, is a partially ionised gas. Some of the atoms and molecules in plasmas would have lost at least one electron, that is, they are ionised.
- The fraction of such atoms need not be high. Degree of ionisation could be fraction of a percent. Thus a plasma would have ions, electrons, neutral atoms and molecules.
- These would be moving around randomly colliding with each other which may result in emission of light, creation of excited particles and so on. Any solid object introduced into the plasma, including the walls of the container, is continuously bombarded by plasma particles.
- Electrons and ions hitting the solid object also create an electric field near the surface. This increases the energy of the bombarding ions. Depending on the type of particle and its energy the surface properties of the material exposed to the plasma can get modified in different ways. Plasma nitriding, surface coating, increasing of surface energy to achieve a hydrophilic surfaces are some examples.
- Plasmas come in different varieties. A major classification distinguishes between non equilibrium low pressure plasmas and high pressure equilibrium plasmas. In non-equilibrium plasmas the electrons and ions would have different temperatures or energies. Collisions are few in such low pressure plasmas and generally electrons are more energetic than ions and other species.
- The electron temperatures could be pretty high  $\sim 10^5$  K. But the overall energy content is low. Non equilibrium plasmas are found in fluorescent lamps, sodium vapour lamps etc. On the other hand, equilibrium plasmas all species i.e. ions, electrons and neutrals are at the same temperature which could be between 5000oK to 10,000oK.
- As the pressures are high total energy content would be high. These are essentially localized sources of high power. They are used electric welding, plasma cutting etc.



- Non equilibrium plasmas are more interesting. Their interaction with matter is species and energy specific. Such plasmas can be tailored to achieve a variety of properties.

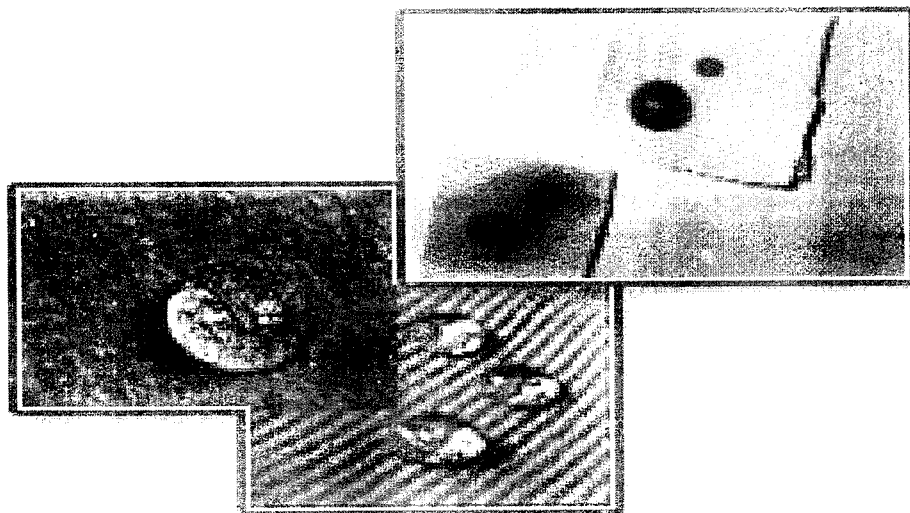


Fig 2.6

- The surface properties alterations obtained by a plasma treatment are complex. Particle induced reactions take usually place in the upper ten nanometers of a surface.
- Short wavelength UV-radiation as it is emitted by low pressure plasmas initiates reactions in a thicker layer (about 100 nm). The relation between the two and the extent of both can be controlled by the process gas and other process parameters.
- The outermost surface, only some atom layers, sometimes less than 1 nm, determines the interaction with other media.
- The chemical composition of this part of a fiber is responsible for good or bad adhesion in lamination or whether the fabric is suitable for impregnation or not.
- Exactly this part of a fiber can be modified by a plasma. For the success of the treatment not only the process parameters contribute, but also the original surface is crucial. Trace amounts of sizings, for example, can modify the reaction condition substantially and have to be taken into consideration in almost every process.

## 2.5.2 Technology

Many different types of treatments have been done so far in batch experiments. This is a legitimate way to demonstrate that a certain treatment can create a desired effect. However, for an industrial application a continuous process is inevitable. There are basically two types of vacuum equipment to realize a continuous web treatment: In an air-to-air system the material passes several differential pumping stages before it reaches the actual process chamber and after it has left it. This allows a real continuous treatment and is particularly suitable if one material is treated in one process for a long period of time without changes.

The metalization of polymer films is done with this technology. For the treatment of different materials, a semi-continuous or roll-to-roll technique is usually more reliable. Deviations of the material properties as for example thickness and water content can be controlled easier than in a continuous treatment. In particular for technological investigations this is the way to go.

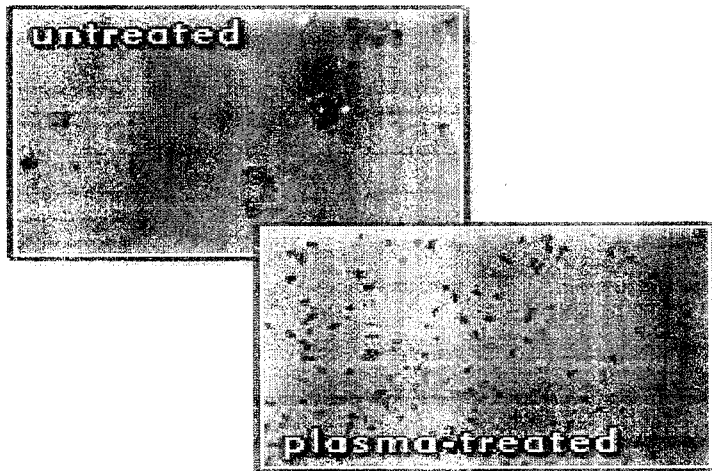


Fig 2.7

### 2.5.3 Processing Methods

➤ **Oxygen or air plasma**

- Removes organic contaminants by chemical reaction with highly reactive oxygen radicals and physical ablation by energetic oxygen ions
- Promotes surface oxidation and hydroxylation (OH groups); increase surface wettability
- Oxidation may be undesirable for some materials (e.g. gold) and can affect surface properties

➤ **Argon plasma**

- Cleans by ion bombardment and physical ablation of contaminants off the surface
- Does not react with the surface or alter surface chemistry

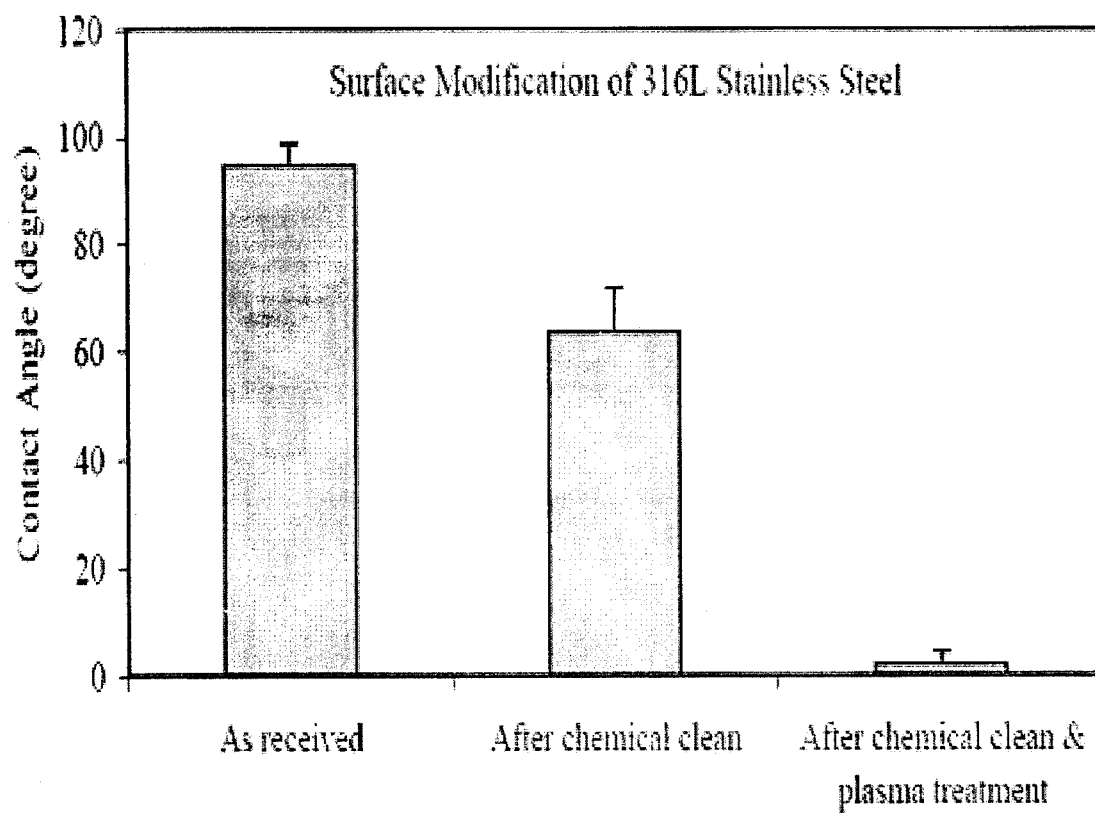
➤ **Carbon Tetra Fluoride (CF<sub>4</sub>) plasma**

- Forms hydrophobic coating of fluorine-containing groups (CF, CF<sub>2</sub>, CF<sub>3</sub>)
- Decreases number of hydrophilic polar end groups on surface; decreases surface wettability.

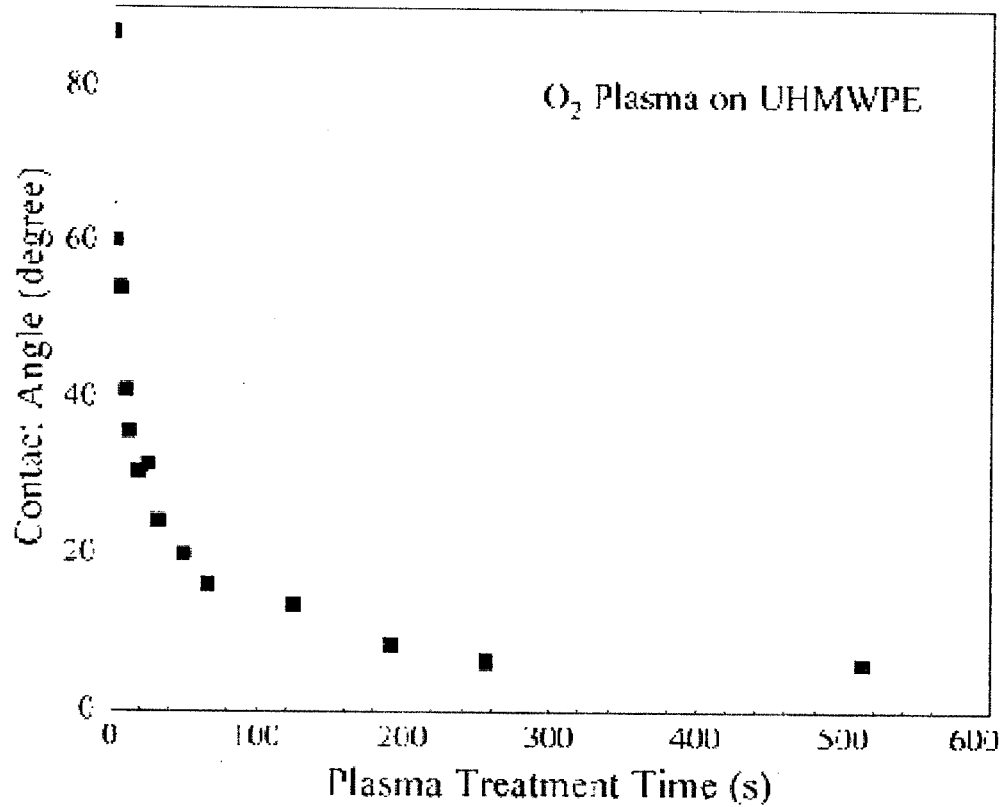
➤ Surfaces should be used immediately after plasma treatment; plasma-treated surfaces may recover the untreated surface characteristics with prolonged exposure to air.

➤ Suggested process parameters values for plasma treatment using a Harrick Plasma cleaner (some experimentation may be required to determine optimal process conditions).

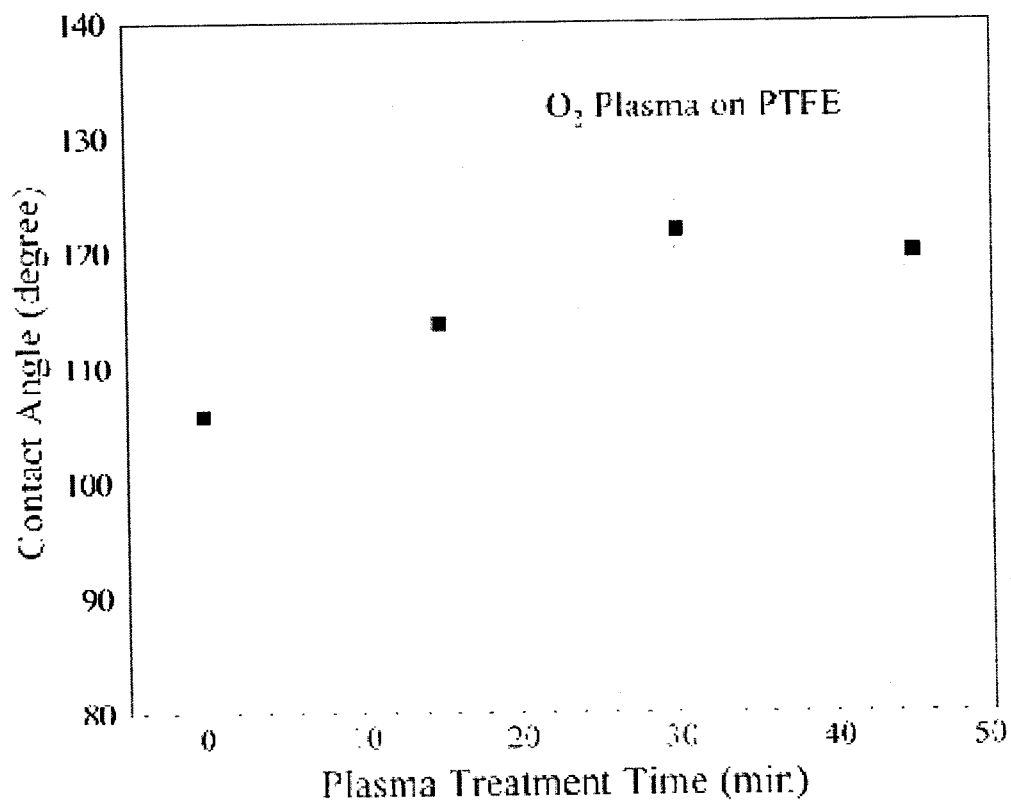
- Pressure: 100 mTorr to 1 Torr
- RF power: Medium or High
- Process time: 1-3 minutes



**Figure 1.** Water drop contact angle measurement on 316L stainless steel (a) as received, (b) after chemical clean (ultrasonication in 70% ethanol, acetone, and 40% nitric acid), and (c) after chemical clean and plasma treatment using a Harrick Plasma cleaner.



**Figure 2.** Water droplet contact angle measurement on ultrahigh molecular weight polyethylene (UHMWPE) as a function of O<sub>2</sub> plasma treatment time using a Harrick Plasma cleaner.



**Figure 3.** Water droplet contact angle as a function of O<sub>2</sub> plasma treatment time, using a Harrick Plasma cleaner, on poly(tetrafluoroethylene) (PTFE), indicating increased hydrophobicity. Plasma treatment produces nanoscale roughness that increases hydrophobicity.

#### 2.5.4 Benefits of Plasma Treatment

- Remove residual organic impurities and weakly bound organic contamination
- Prepare surfaces for subsequent processing (e.g. film deposition or adsorption of molecules)
- Improve surface coverage and spreading of coatings and enhance adhesion between two surfaces
- Modify wettability to render a surface hydrophilic or hydrophobic with the appropriate process gas(es)
- Affect only a few monolayers of the surface; does not change bulk properties of the material
- Can treat a wide variety of materials as well as complex surface geometries; examples include:
  - Semiconductor wafers and substrates (Si, Ge)
  - Glass slides and substrates.
  - Oxides (quartz, indium tin oxide (ITO), TiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>); mica

### **2.5.5 Applications**

- Surface preparation of substrates prior to self-assembly experiments
- Surface preparation of electron microscopy (EM) sample grids
- Plasma cleaning of printed circuit (PC) boards and die surfaces prior to bonding
- Plasma treatment of dental implant and impression mold materials
- Plasma treatment of biomaterials and biomedical devices prior to functionalizing surface
- Plasma treatment of fibers to improve adhesion to matrix in fiber-reinforced composite materials
- Study of adhesion characteristics of dissimilar materials by mechanical testing or atomic force microscopy (AFM) force measurements

### **2.5.6 EFFECT OF PLASMA TREATMENT ON SOME PROPERTIES OF COTTON**

- Treatment of cotton fibres has been studied in air and oxygen plasma and the treatment time, nature and flow rate of the gas, and plasma power have been varied. In order to establish the chemical effect of plasma treatment on cotton fibres the following tests have been performed:
- Cuprammonium fluidity test, weight loss measurement, determination of carboxyl groups, carbonyl group identification, FTIR analysis and measurement of the ASTM yellowness of the untreated and plasma treated cotton fabrics. In addition, vertical wicking studies and the effect of ageing of the plasma treated samples on the rate and the amount of dye uptake have been investigated.
- The plasma treatments lead to surface erosion of the cotton fibres which generates a weight loss accompanied by an increase in the fibre carboxyl group and carbonyl group contents.
- The increase in fibre carboxyl group content leads to a more wettable fibre and the rate of fabric vertical wicking is increased.
- The direct dye (chloramine Fast Red K) uptake of treated samples increases almost linearly with the increase in fibre carboxyl group content caused by plasma treatment, but progressively decreases with increase in the ageing time after oxygen plasma treatment. Ageing after plasma treatment also increases the fabric yellowness.

## 2.6 EXTRACTION METHODS

Two methods:

### 1. Cold Method.

- Shaking Method

### 2. Hot Method

- Soxhlet Apparatus

#### 1. Shake method

The extract is taken in a flask, mixed with water and is shaken continuously for 2-3 days.

#### 2. Soxhlet Apparatus

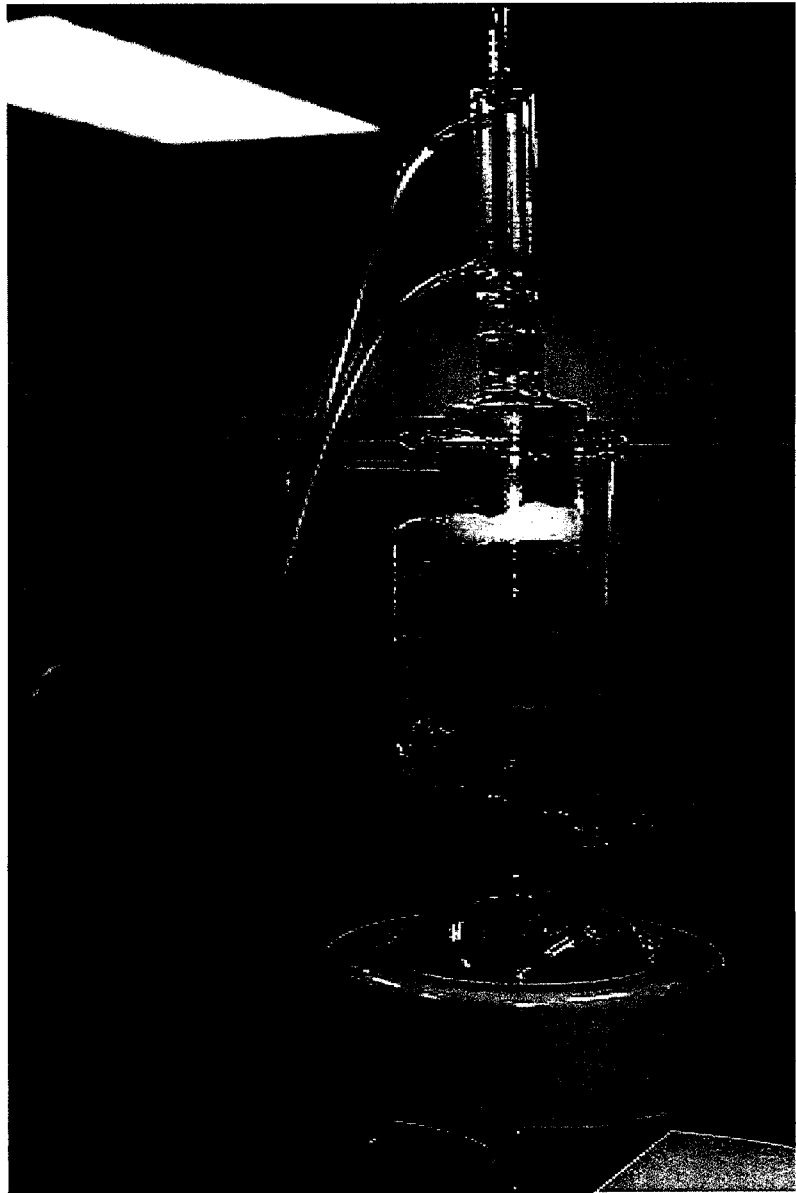
Soxhlet extraction is only required where the desired compound has a limited solubility in a solvent, and the impurity is insoluble in that solvent. If the desired compound has a significant solubility in a solvent then simple filtration is sufficient. Fruit extraction in progress. The sample is placed in the thimble.

Normally a solid material containing some of the desired compound is placed inside a thimble made from thick filter paper, which is loaded into the main chamber of the Soxhlet extractor.

The Soxhlet extractor is placed onto a flask containing the extraction solvent. The Soxhlet is then equipped with a condenser.

The solvent is heated to reflux. The solvent vapour travels up a distillation arm, and floods into the chamber housing the thimble of solid.





**Fig 2.9**

The condenser ensures that any solvent vapour cools, and drips back down into the chamber housing the solid material. During each cycle, a portion of the non-volatile compound dissolves in the solvent. After many cycles the desired compound is concentrated in the distillation flask.

The advantage of this system is that instead of many portions of warm solvent being passed through the sample, just one batch of solvent is recycled. After extraction the solvent is removed, typically by means of a rotary evaporator, yielding the extracted compound.

The non-soluble portion of the extracted solid remains in the thimble, and is usually discarded.

The chamber containing the solid material slowly fills with warm solvent. Some of the desired compound will then dissolve in the warm solvent.

When the Soxhlet chamber is almost full, the chamber is automatically emptied by a siphon side arm with the solvent running back down to the distillation flask. This cycle may be allowed to repeat many times over hours or days.

## **2.7 FABRIC TREATMENT WITH HERBAL PRODUCTS**

### **2.7.1. Direct Application Method**

Ethanol extracts of the herbs were directly applied on 100% cotton fabric by pad-dry-cure method. 1% of the herbal extract was applied on cotton fabric along with 6% citric acid as cross-linking agent by pad-dry-cure method. Padding was carried out in pneumatic padding mangle at a pressure of 3 psi to get a pick up of 100% on weight of fabric. Drying and curing were carried out in spooner at 80 °C for 5 min and 150 °C for 3 min respectively.

### **2.7.2. Micro-encapsulation Method**

Micro-encapsulation is one of the novel methods of getting functional finishes on textiles. Its a process by which very tiny droplets of liquid or particles of solid are covered with a continuous film of polymer material. Micro-encapsulation is more advantageous to conventional processes in terms of economy, energy saving, ecofriendliness and controlled release of substance. The agents reside in colloidal suspension within the amorphous zone of the binder so that a reservoir of agent is present in solid solution within the polymer matrix. Such treated fabrics were reported to be durable to a few number of wash cycles. The prolonged bioactivity of the fabric is due to slow diffusion of the antimicrobial agent out of the polymer reservoir. In the present work, micro-encapsulation was done using herbal extract as core material and gum acacia as wall materials. Ten grams of acacia powder was allowed to swell for 15 min in 100 ml of hot water. To this mixture, 50 ml of hot water was added and stirred for 15 min maintaining the temperature between 40 c and 60 c. One and a half gram of core material (herbal extract) was slowly added under stirring condition. Stirring was continued for another 15 min and then 10 ml of 20% sodium sulphate and 6 g of citric acid were added. The stirring was stopped and the mixture was freeze dried in a freezer to develop microcapsules solution and padded through pneumatic padding mangle at a pressure of 3 psi. The treated fabric was dried at 80 c for 5 min.

### **2.7.3 Cross-Linking method**

One gram of herbal extract was mixed with 100ml (120 gpl conc.) non-formaldehyde based resin (KRISOF NIL/F) and 2.0g MgCl<sub>2</sub> was added as a catalyst. Cotton fabric was dipped in the resin solution and padded through a pneumatic padding mangle. The treated fabric was dried at 80 c for 5 min and cured at 150 c for 3 min.

## 2.8. TESTING METHODS

### 2.8.1 ANTIMICROBIAL PROPERTIES

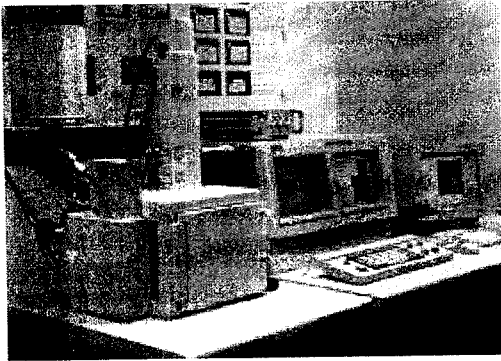
#### **Enhancing the Antibacterial Properties of Cotton Fabric**

Attempts were made to enhance the antibacterial properties of the cotton fabrics via pre-crosslinking with tri-methylol melamine followed by subsequent treatment with iodine (NH and NI) solution to create new active sites. The treated fabric shows the ability to inhibit as well as to arrest the growth of both of the bacillus subtilis and Escherichia coli. The antibacterial activity is determined by the degree of modification, the extent of iodine retention, as well as the ease of iodine liberation. Repeated laundering-recharging cycle could fully bring back the antibacterial functions of the finished cotton fabrics.

#### **Durable Press and Antimicrobial Finishing of Cotton Fabrics with a Citric Acid and Chitosan Treatment**

Citric acid (cA) and chitosan are used as durable press and antimicrobial finishing agents for cotton and are applied by means of the conventional pad-dry-cure process. CA is expected to react with hydroxyl groups in cellulose and chitosan or with amino groups in chitosan to form ester crosslinking or an inter-ionic attraction. Durable press appearance ratings of 3.5-4 are imparted with CA and chitosan treatments. The cotton fabric treated with CA alone shows antimicrobial properties. These and durable press performance are retained through twenty washing and tumble drying cycles. Strength retention improves more with CA and chitosan than with CA alone.

## 2.8.2 SCANNING ELECTRON MICROSCOPE (SEM)



A typical SEM instrument, showing the electron column, sample chamber, EDS detector, electronics console, and visual display monitors.

The scanning electron microscope (SEM) uses a focused beam of high-energy electrons to generate a variety of signals at the surface of

Fig 2.10

solid specimens. The signals that derive from electron-sample interactions reveal information about the sample including external morphology (texture), chemical composition, and crystalline structure and orientation of materials making up the sample. In most applications, data are collected over a selected area of the surface of the sample, and a 2-dimensional image is generated that displays spatial variations in the properties

. Areas ranging from approximately 1 cm to 5 microns in width can be imaged in a scanning mode using conventional SEM techniques (magnification ranging from 20X to approximately 30,000X, spatial resolution of 50 to 100 nm). The SEM is also capable of performing analyses of selected point locations on the sample; this approach is especially useful in qualitatively or semi-quantitatively determining chemical compositions (using EDS), crystalline structure, and crystal orientations (using EBS).

The design and function of the SEM is very similar to the EPMA and considerable overlap in capabilities exists between the two instruments.

### **Fundamental Principles of Scanning Electron Microscopy (SEM)**

Accelerated electrons in an SEM carry significant amounts of kinetic energy, and this energy is dissipated as a variety of signals produced by electron-sample interactions when the incident electrons are decelerated in the solid sample. These signals include secondary electrons (that produce SEM images), backscattered electrons (BSE), diffracted backscattered electrons (EBS) that are used to determine crystal

structures and orientations of minerals), photons (characteristic X-rays that are used for elemental analysis and continuum X-rays), visible light (cathodoluminescence--CL), and heat. Secondary electrons and backscattered electrons are commonly used for imaging samples: secondary electrons are most valuable for showing morphology and topography on samples and backscattered electrons are most valuable for illustrating contrasts in composition in multiphase samples (i.e. for rapid phase discrimination). X-ray generation produced by inelastic collisions of the incident electrons with electrons in discrete orbitals (shells) of atoms in the sample. As the excited electrons return to lower energy states, they yield X-rays that are of a fixed wavelength (that is related to the difference in energy levels of electrons in different shells for a given element). Thus, characteristic X-rays are produced for each element in a mineral that is "excited" by the electron beam. SEM analysis is considered to be "non-destructive"; that is, x-rays generated by electron interactions do not lead to volume loss of the sample, so it is possible to analyze the same material repeatedly. **Scanning Electron Microscopy (SEM) Instrumentation - How Does It Work?**

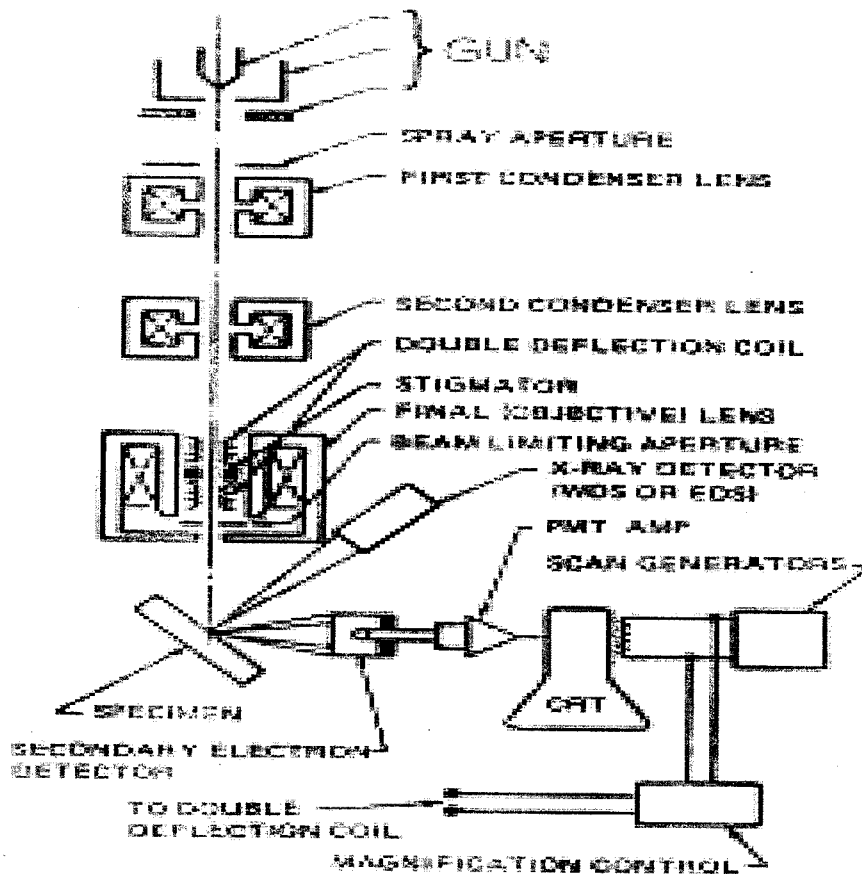


Figure 2.11. Schematic drawing of the electron and x-ray optics of a combined SEM-EDMA.

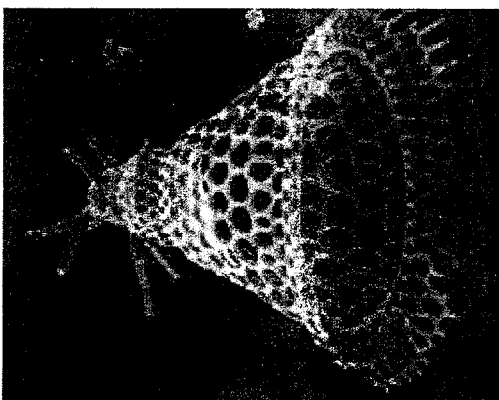
Fig 2.11

Essential components of all SEMs include the following:

- Electron Source ("Gun")
- Electron Lenses
- Sample Stage
- Detectors for all signals of interest
- Display / Data output devices
- Infrastructure Requirements:
  - Power Supply
  - Vacuum System
  - Cooling system
  - Vibration-free floor
  - Room free of ambient magnetic and electric fields

SEMs always have at least one detector (usually a secondary electron detector), and most have additional detectors. The specific capabilities of a particular instrument are critically dependent on which detectors it accommodates.

## Applications



The SEM is routinely used to generate high-resolution images of shapes of objects (SEI) and to show spatial variations in chemical compositions:

- 1) acquiring elemental maps or spot chemical analyses using EDS,
- 2) discrimination of phases based on mean atomic number (commonly related to relative density) using BSE and
- 3) compositional maps based on differences in trace element "activators" (typically transition metal and Rare Earth elements) using CL.

The SEM is also widely used to identify phases based on qualitative chemical analysis and/or crystalline structure. Precise measurement of very small features and objects down to 50 nm in size is also accomplished using the SEM. Backscattered electron images (BSE) can be used for rapid discrimination of phases in multiphase samples. SEMs equipped with diffracted backscattered electron detectors (EBSD) can be used to examine microfabric and crystallographic orientation in many materials.

### Strengths and Limitations of Scanning Electron Microscopy (SEM)

#### **Strengths**

There is arguably no other instrument with the breadth of applications in the study of solid materials that compares with the SEM. The SEM is critical in all fields that require characterization of solid materials. While this contribution is most concerned with geological applications, it is important to note that these applications are a very small subset of the scientific and industrial applications that exist for this instrumentation. Most SEM's are comparatively easy to operate, with user-friendly "intuitive" interfaces. Many applications require minimal sample preparation. For many applications, data acquisition is rapid (less than 5 minutes/image for SEI, BSE, spot EDS analyses.) Modern SEMs generate data in digital formats which are highly portable.



## Limitations

Samples must be solid and they must fit into the microscope chamber. Maximum size in horizontal dimensions is usually on the order of 10 cm, vertical dimensions are generally much more limited and rarely exceed 40 mm. For most instruments samples must be stable in a vacuum on the order of  $10^{-5}$  -  $10^{-6}$  torr. Samples likely to outgas at low pressures (rocks saturated with hydrocarbons, "wet" samples such as coal, organic materials or swelling clays, and samples likely to decrepitate at low pressure) are unsuitable for examination in conventional SEM's.

However, "low vacuum" and "environmental" SEMs also exist, and many of these types of samples can be successfully examined in these specialized instruments. EDS detectors on SEM's cannot detect very light elements (H, He, and Li), and many instruments cannot detect elements with atomic numbers less than 11 (Na). Most SEMs use a solid state x-ray detector (EDS), and while these detectors are very fast and easy to utilize, they have relatively poor energy resolution and sensitivity to elements present in low abundance when compared to wavelength dispersive x-ray detectors (WDS) on most electron probe microanalyzers (EPMA). An electrically conductive coating must be applied to electrically insulating samples for study in conventional SEM's, unless the instrument is capable of operation in a low vacuum mode.

## User's Guide - Sample Collection and Preparation

Sample preparation can be minimal or elaborate for SEM analysis, depending on the nature of the samples and the data required. Minimal preparation includes acquisition of a sample that will fit into the SEM chamber and some accommodation to prevent charge build-up on electrically insulating samples. Most electrically insulating samples are coated with a thin layer of conducting material, commonly carbon, gold, or some other metal or alloy.

The choice of material for conductive coatings depends on the data to be acquired: carbon is most desirable if elemental analysis is a priority, while metal coatings are most effective for high resolution electron imaging applications. Alternatively, an electrically insulating sample can be examined without a conductive coating in an instrument capable of "low vacuum" operation.

**MATERIALS AND METHODOLOGY**

## CHAPTER 3

### 3.1 METHODOLOGY

Herbal Extraction from Plant



Characterization of

-Antimicrobial activity.

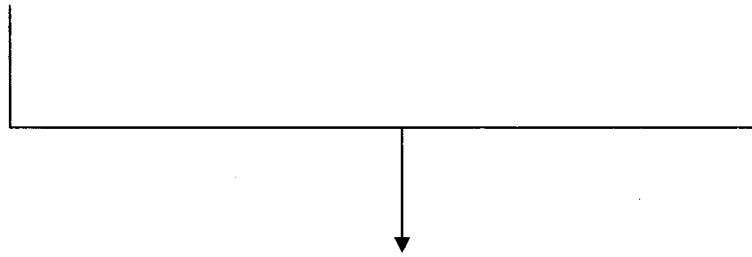
Plasma Treatment of Cotton and  
bamboo



Characterization Of treated fabric

-Tensile property

-Water vapour Permeability



Application of herbal extract on plasma

Treated material



Characterization of

-antimicrobial activity

-Effect of plasma treatment on

-wound healing activity.

-compare both the material characteristics.

### **3.2 APPLICATION OF EXTRACT FROM TULSI POWDER:**

Active substances were extracted from the tulasi powder using ethanol-water (1:1) mixture. This mixture was arrived at after attempting on various solvents and their mixtures for extraction purpose. Solvent extraction technique using soxhlet apparatus was employed for the purpose.

### **3.3 PROCEDURE ADOPTED FOR HERBAL EXTRACTION:**

For every 8g of powder 80ml of ethanol-water mixture was used in the process. Tulsi powder was taken in the thimble and ethanol-water mixture was adopted to the extraction flask. The complete assembly of the apparatus was kept on a heating mantle and its temperature was adjusted in such a way that the solvent siphons over once in every 20 min interval. The extraction was continued for 8 hours. After turning off the heat, the thimble was replaced with a condenser and heating continued till the volume of the solution in the flask was reduced to 20ml. Then the solution in the flask was transferred to a beaker and heating mantle to evaporate the ethanol-water mixture. The residual extract was further converted into powder form by applying vacuum in vacuum desiccators.

### **3.4 APPLICATION OF HERBAL EXTRACT**

Ethanol extracts of the herbs were directly applied on 100% cotton fabric and bamboo fabric by **pad-dry-cure** method. 1% of the herbal extract was applied on cotton fabric and bamboo fabric along with 6% citric acid as cross-linking agent by pad-dry-cure method. Padding was carried out in a pneumatic padding mangle at a pressure of 3 psi to get a pick up of 100% on weight of fabric.

Drying- 80 c for 5 min

Curing- 120 c for 3 min

The samples were dried at ambient temperature without washing. The quantity of herbal extract present in the fabric after drying was found to be 4.57% owm and 6.66% owm. Then the treated (unwashed) samples were washed using the procedure described in ISO:6330-1984E test method using

3%(v/v) soap solution at 40 c for 30 min using l:m 50:1 repeatedly for four times and dried. Both the treated and washed samples were taken for antibacterial testing.

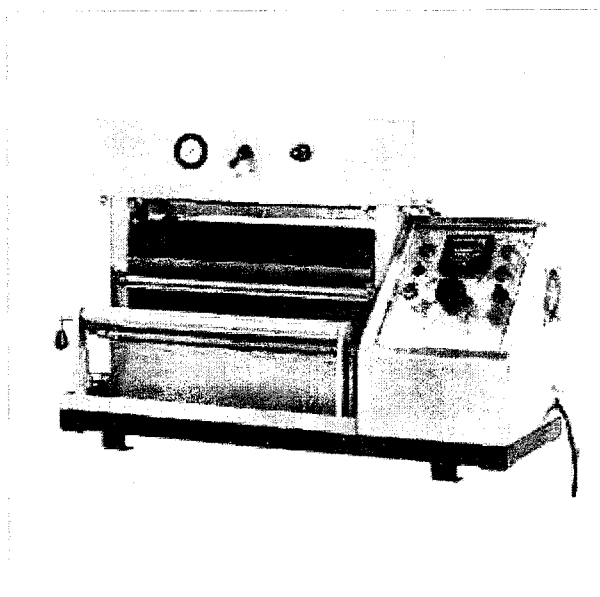


Fig 3.1

### 3.5 ESTIMATION OF ANTI-MICROBIAL ACTIVITY

Testing for antimicrobial activity of treated and washed samples was carried out using a qualitative method as per the standard AATCC test method 147(Disc diffusion method).Mueller Hinton agar solution was prepared by dissolving Mueller Hinton agar solution(38 g/l) and nutrient broth (13 g/l) in distilled water(100 ml) taken in a conical flask. Then the flask was autoclaved at 121 c for 20 min. Sterilized petri plates were prepared with an equal thickness of Mueller Hinton agar medium. The incubated culture was swabbed on the agar plates using swab stick. Disc shaped treated samples of diameter 1.8 cm were placed on the seeded agar plate. After overnight incubation, the zones of inhibition were measured using the following formula:

$$\text{Average width of zone of Inhibition, (Z.O.I) = (T-D)/2}$$

Where,

T is the total diameter of the test specimen and clear zone (mm) and

D is the diameter of the test specimen (mm).

### **3.6 PLASMA TREATMENT**

To treat the cotton and bamboo fabrics with plasma. There are different methods of plasma treatment available. In this project, we selected oxygen or air plasma method. Plasma treatment is applied to cotton and bamboo separately.

#### **Oxygen or air plasma**

- Removes organic contaminants by chemical reaction with highly reactive oxygen radicals and ablation by energetic oxygen ions
- Promotes surface oxidation and hydroxylation (OH groups); increase surface wettability
- Oxidation may be undesirable for some materials (e.g. gold) and can affect surface properties.

#### **PARTICULARS:**

Distance=5cm

Time=5 mins

Voltage=500 volt

**RESULTS AND DISCUSSIONS**



# THE SOUTH INDIA TEXTILE RESEARCH ASSOCIATION

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XXIV/ 2561 /2012

06.02.2012

Mr.M.Mohamed Ashif,  
Kumaraguru College of Technology,  
Post Box No. 2034,  
Coimbatore – 641 006.

**Ch.T.R.No.2909**

Dear Sir,

This is with reference to your letter dated 30.01.2012 regarding Tulasi powder sample given to us for testing and the results are as follows:

### Assessment of Antibacterial Activity as per Kirby Bauer Susceptibility Test

Sample No.	Sample Particulars	Zone of Bacteriostasis (in mm)		
		<i>Staphylococcus aureus</i> (ATCC 6538)	<i>Klebsiella pneumoniae</i> (ATCC 4352)	Remarks
CM5962	Tulasi Powder Sample	15	13	Good Antibacterial Activity

Inoculum size : *S.aureus* :  $1.5 \times 10^8$  cfu/ml, *K.pneumoniae* :  $1.5 \times 10^8$  cfu/ml

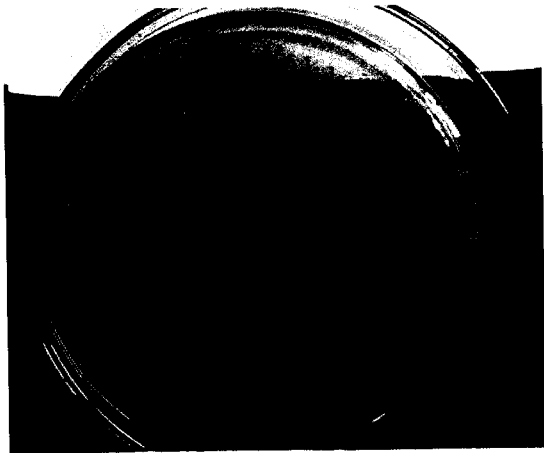
Yours faithfully,

Incharge-Testing(Textile Chemistry)

Encl : Bill



**PLASMA TREATED SAMPLE**



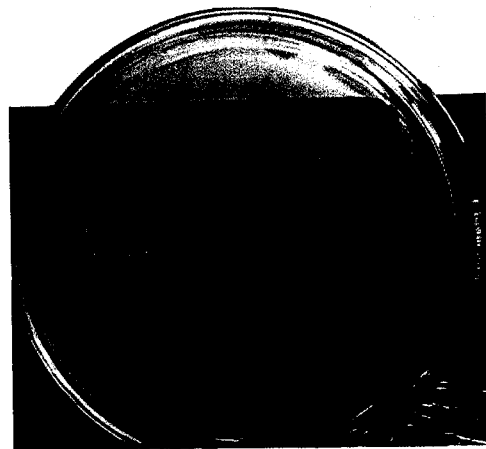
**Cotton Conc. 10**



**Cotton Conc. 30**



**Bamboo Conc. 10**

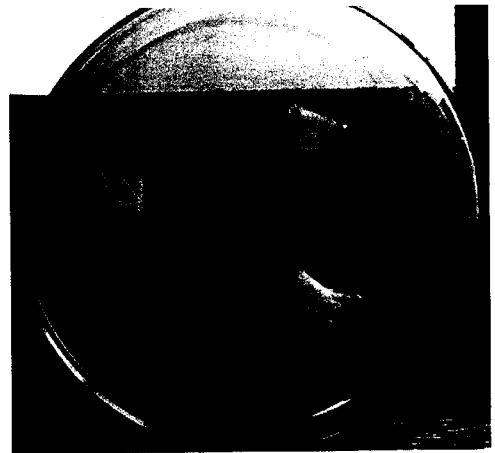


**Bamboo Conc. 30**

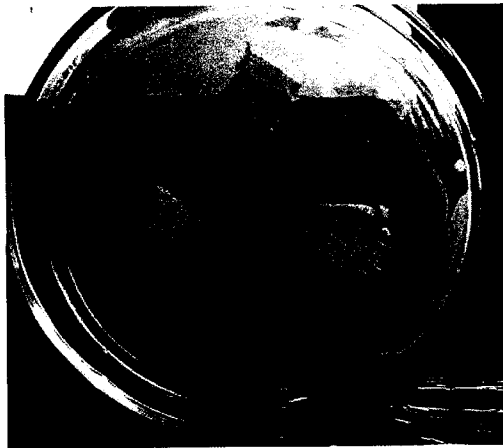
**PLASMA UN-TREATED SAMPLE**



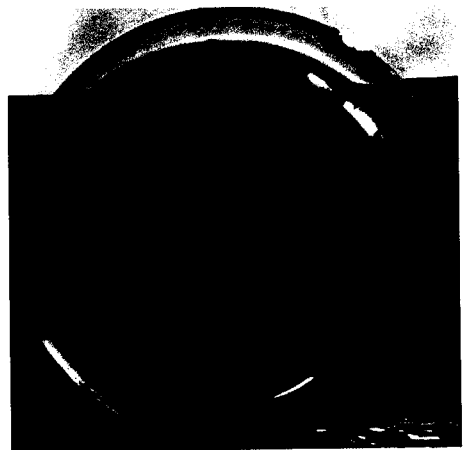
**Cotton Conc. 10**



**Cotton Conc. 30**

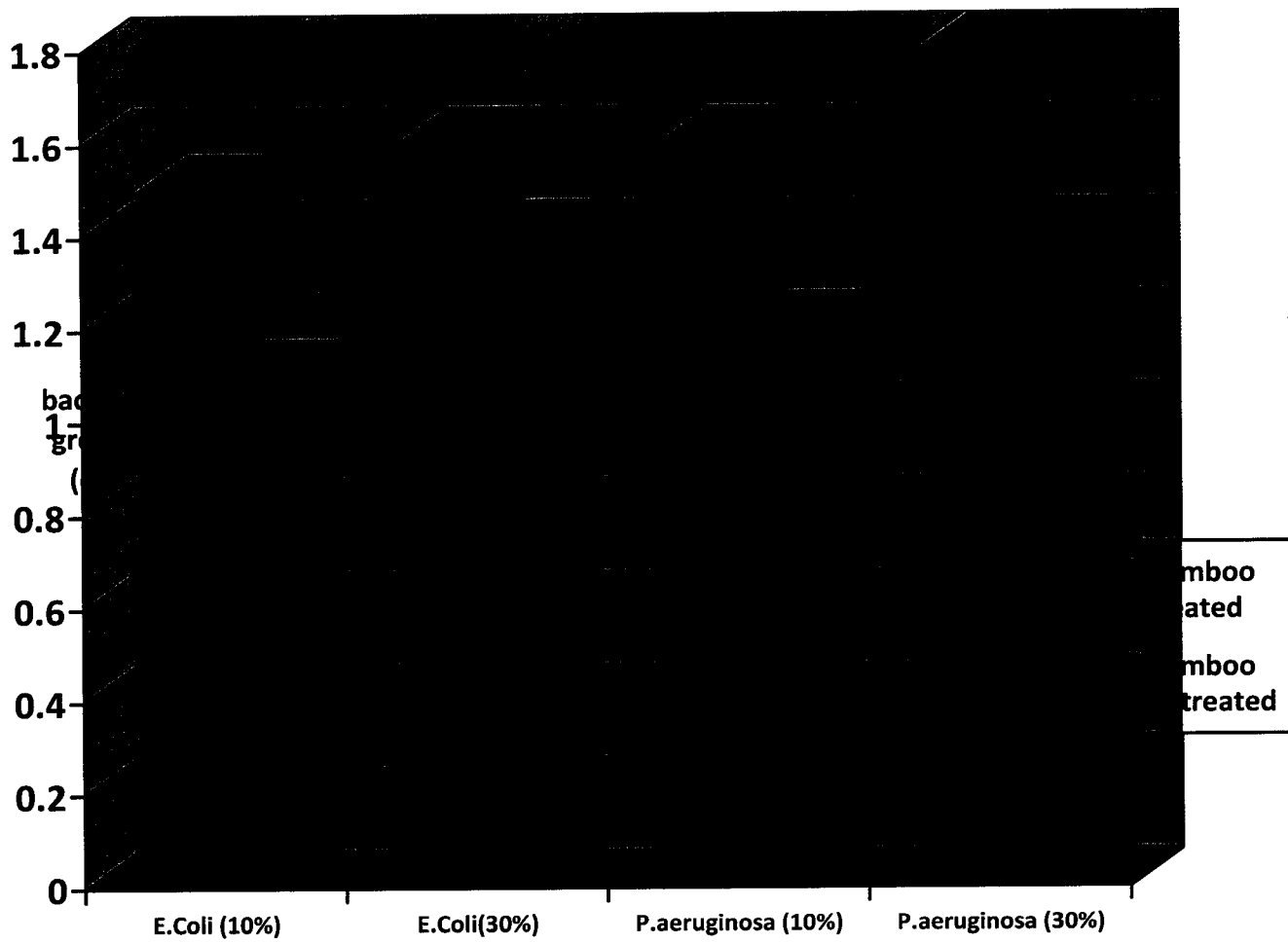


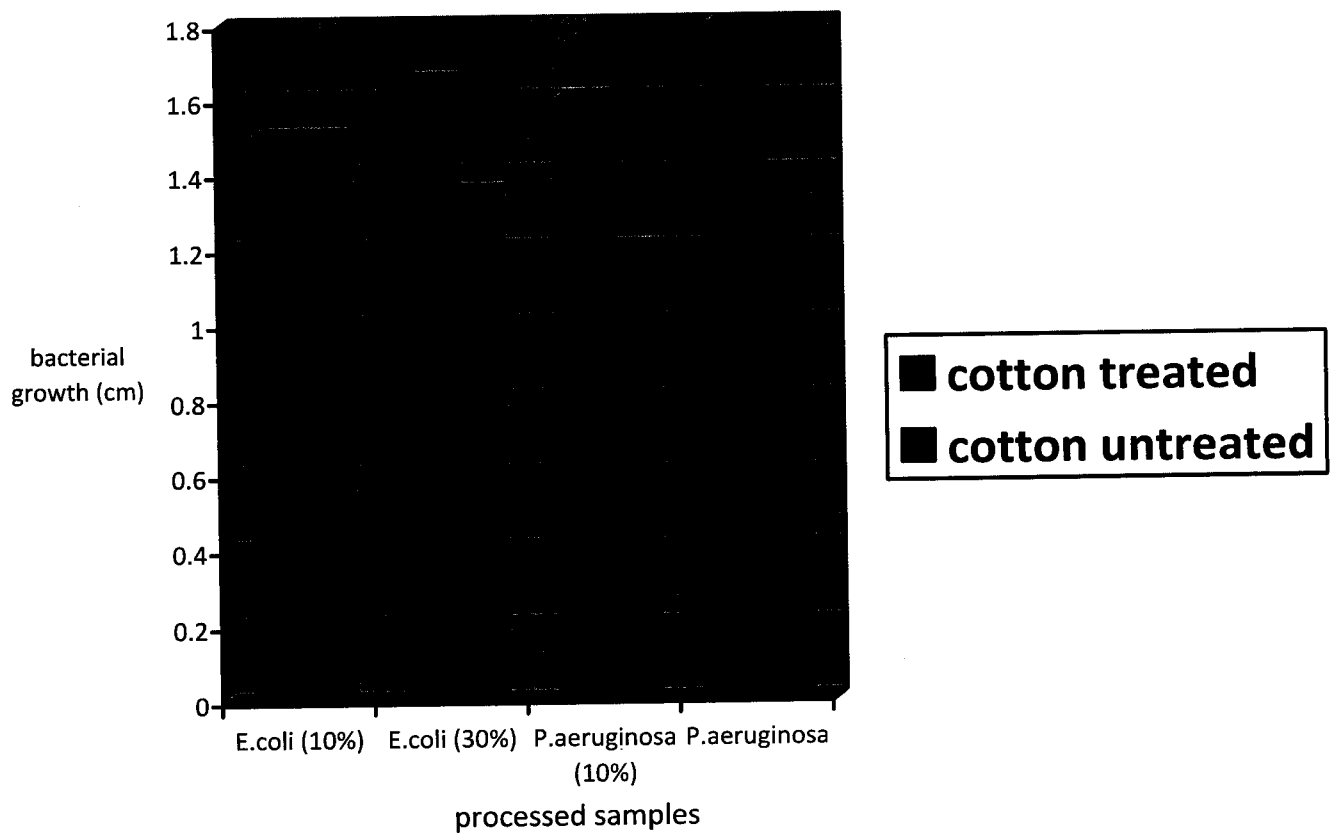
**Bamboo Conc. 10**



**Bamboo Conc. 30**

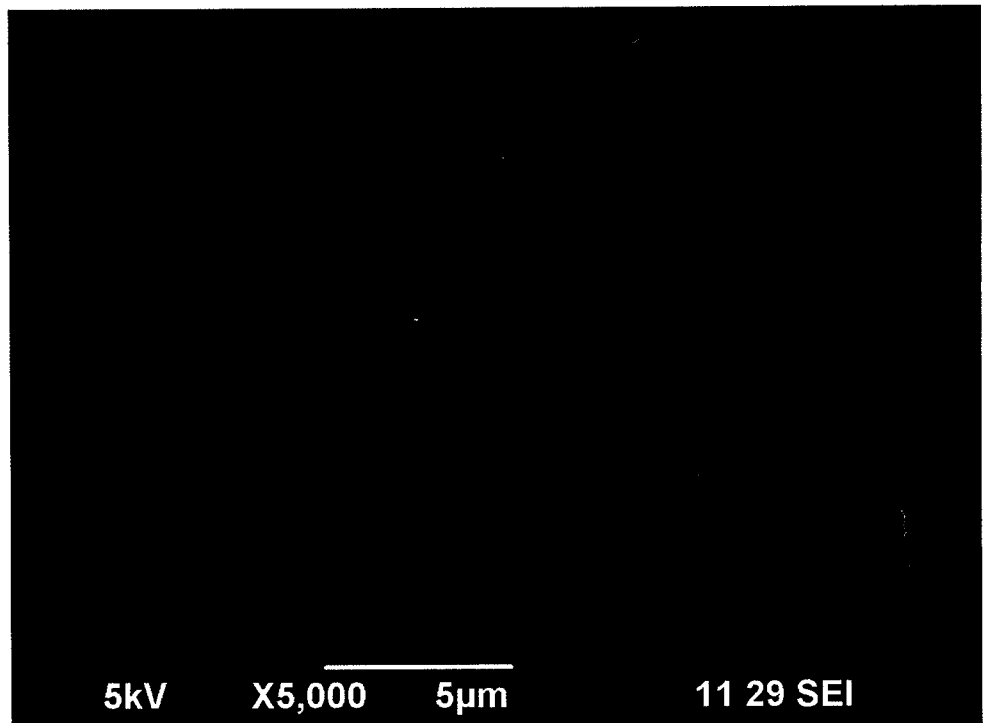
Materials	Concentration (%)	Zone of inhibition in diameter(cm)	
		Microbial strains tested	
		E.Coli	P.aeruginosa
Cotton treated	10	1.5	1.6
	30	1.65	1.8
Cotton untreated	10	1.5	1.2
	30	1.35	1.4
Bamboo treated	10	1.5	1.8
	30	1.6	2.0
Bamboo untreated	10	1.7	1.2
	30	1.5	1.4



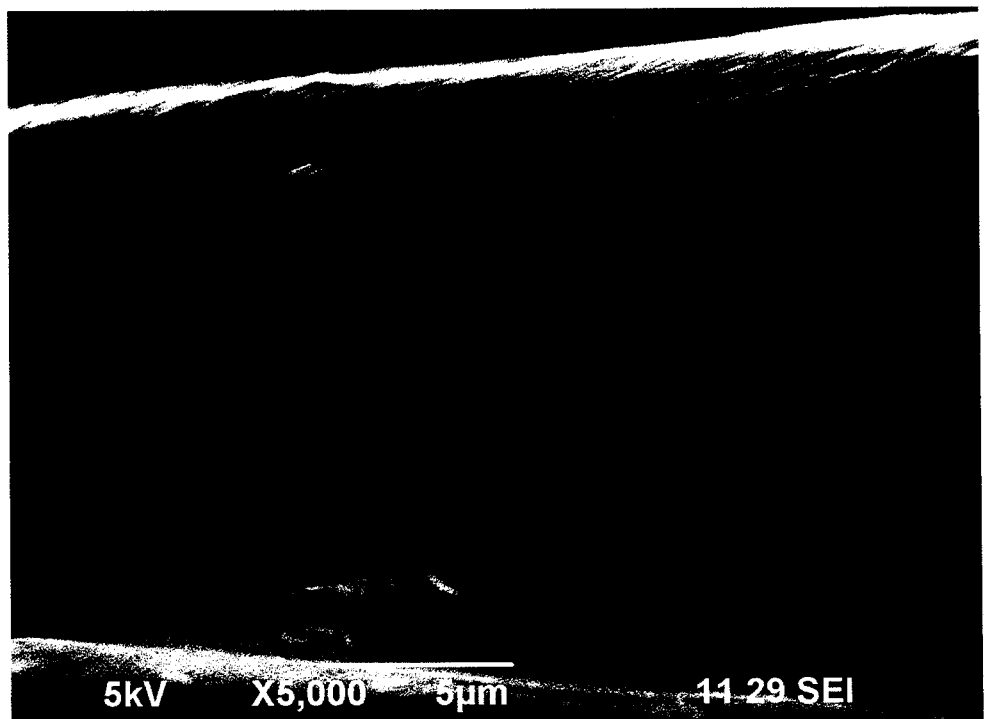


# SCANNING ELECTRON MICROSCOPE(SEM)

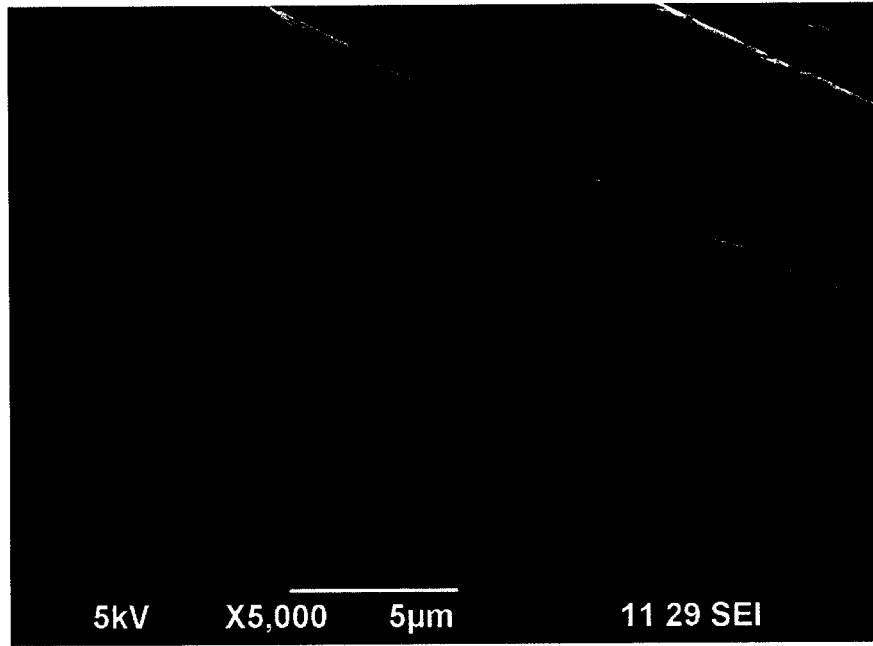
Cotton Treated



Cotton Untreated



**Bamboo Treated**



**Bamboo untreated**



## **Conclusions**

The following conclusions were made from the work done.

1. It is proved that Tulsi has antimicrobial activity in its liquid extract form.
2. Plasma treatment enhanced the antimicrobial activity on both cotton and bamboo fibres.
3. The concentration of the herbal extract played a vital role. The fabric treated with conc.30 ha a good antimicrobial activity.
4. Compared to cotton, bamboo has got higher antimicrobial property.



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