

B.TECH DEGREE EXAMINATIONS: OCTOBER/NOVEMBER-2008

Third Semester

BIOTECHNOLOGY**U07BT308 - Principles of Chemical Engineering**

Time: Three Hours

Maximum Marks: 100

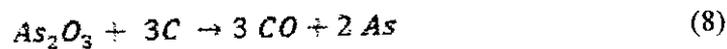
Answer ALL Questions:-

Part-A (20x1 = 20 Marks)

- The temperature of a chemical process was found to be 80°C. What is the temperature in °F?
a) 353°F b) 636°F c) 176°F d) -193°F
- Which one of the following is a derived unit in SI system?
a) Mass b) Volume c) Time d) Length
- The area of an irregular surface can be determined by using
a) Newton's Forward Difference Formula
b) Newton's Backward Difference Formula
c) Simpson's 1/3rd rule
d) Numerical Differentiation
- In which of the following reactions, Law of conservation of mass becomes invalid?
a) Ammonia synthesis b) Fermentation of Ethanol
c) Combustion reactions d) Nuclear Fission reactions
- A process is said to be at steady state if its conditions
a) Do not change with respect to position
b) Change with respect to position
c) Change with respect time
d) Do not Change with respect to time
- An aqueous solution of NaCl has a concentration of 24 wt% NaCl with a density of 1.178 g/cm³ at 25°C. The mole fraction of water in the solution is
a) 0.0886 b) 0.9114 c) 4.22 d) 0.41
- The stream which skips one or more intermediate stages of a process is known as
a) Purge stream b) Bypass stream c) Recycle stream d) None of these
- Steady state temperature reached by a small amount of liquid evaporating into a large amount of unsaturated vapor-gas mixture is\
a) Dry-bulb temperature b) Wet-bulb temperature
c) Dew point d) Adiabatic saturation temperature
- Second law of thermodynamics is concerned with
a) Amount of energy transferred b) Direction of energy transfer
c) Irreversible processes only d) Non-cyclic processes only
- Which of the following is a thermodynamic property of a system?
a) Concentration b) Mass c) Temperature d) Vapour pressure
- The quantity of heat that can not be measured by using a temperature measuring device is known as
a) Latent heat b) Sensible heat c) Specific heat d) Preheat
- When a solute is added to a solvent, its vapour pressure
a) Increases b) Decreases
c) is not affected d) Initially increases but then start to decrease.

b) i) Define the following: Mole, Molarity, Percentage Humidity and Saturation Humidity. (4x2=8)

ii) For the following reaction, how many grams of charcoal is required to reduce 1500 kg of arsenic trioxide? (Atomic weight of Arsenic = 75)



23. a) Write short notes on Clasius-Clapeyron equation, Watson's equation, Antoine equation, and Riedel equation. (4x4=16)

(OR)

b) A liquid fermentation medium at 30° C is pumped at a rate of 2000 kg/h through a heater, where it is heated to 70° C under pressure. The waste heat water used to heat this medium enters at 95° C and leaves at 85° C. The average heat capacity of the fermentation medium is 4.06 kJ/kg.K. and that for water is 4.21 ° kJ/kg.K. The fermentation medium and the waste water stream are separated by a metal surface through which heat is transferred and do not physically mix with each other. Make a complete heat balance on the system. Calculate the water flow and the amount of heat added to the fermentation medium assuming no heat losses. (16)

24. a) i) Give Newton's law of viscosity and explain various types of fluids based on this law. (10)

ii) Explain Briefly about Reynold's Number and its significance in Fluid flow. (6)

(OR)

b) i) A liquid having a density of 801 kg/m³ and a viscosity of 1.49 x 10⁻³ Pa.S is flowing through a horizontal straight pipe at a velocity of 4.57 m/s. The diameter of the pipe is 50 mm. For a length of pipe of 61m, determine the type of flow and the frictional head loss. (10)

ii) State Pascal's law and its applications in Chemical engineering. (6)

25. a) What is fluidization? Explain and the behavior of a fluidized bed and its applications in detail. (16)

(OR)

b) i) Explain the working of a Centrifugal pump with a neat sketch (10)

ii) Explain briefly about Characteristic and operating curves of a Centrifugal pump (6)
