



M.TECH DEGREE EXAMINATIONS: NOV/DEC 2023

(Regulation 2018)

First Semester

BIOTECHNOLOGY

P18BTI1202: Bioprocess Modelling and simulation

COURSE OUTCOMES

CO1: Conceptualize mathematical and engineering concepts in bioprocess modeling and simulation.

CO2: Identify and analyze mathematical model in biochemical engineering systems.

CO3: Select the appropriate components of SuperPro Design software.

CO4: Apply the concepts of MATLAB and SIMULINK in bioprocess systems.

CO5: Ability to solve and analyze data using MATLAB.

CO6: Apply, design and interpret process flowsheeting using SuperPro Design software.

Time: Three Hours

Maximum Marks: 100

Answer all the Questions:-

PART A (10 x 1 = 10 Marks)

1. Choose the correct order from the options below: CO1 [K₂]
 - a) Model hypothesis, Experimental design, Perform experiments, Evaluation of experiments
 - b) Experimental design, Perform experiments, Evaluation of experiments, Model hypothesis
 - c) Perform experiments, Experimental design, Model hypothesis, Evaluation of experiments
 - d) Model hypothesis, Evaluation of experiments, Perform experiments, Experimental design
2. Which term describes a state in which no changes occur over time, indicating a balanced condition in a bioprocess system? CO1 [K₂]
 - a) Steady State
 - b) Transient State
 - c) Dynamic State
 - d) Equilibrium State
3. What is the main objective of utilizing flowsheets in chemical engineering processes? CO2 [K₂]
 - a) Scheduling
 - b) Throughput analysis
 - c) Economic evaluation
 - d) Equipment utilization analysis
4. Match list I (Organism) with list II and select the correct answer (carbon content) using the codes given below the list: CO3 [K₃]

MATLAB for problem-solving?

- a) Symbolic computation
- b) Matrix factorization
- c) Approximating solutions
- d) Geometric transformations

PART B (10 x 2 = 20 Marks)

- 11. How do chemical kinetics is utilized to model and predict microbial growth in a bioprocess. CO1 [K₃]
- 12. Mention any two role of transport equations in modeling the movement of nutrients and metabolites within a bioprocess system. CO1 [K₃]
- 13. Enlist any two advantages and disadvantages of using an enclosed vessel in a biochemical engineering system compared to an open vessel. CO2 [K₂]
- 14. Illustrate how a continuous flow tank differs from a steam jacketed vessel in biochemical engineering systems, emphasizing their respective functions and applications. CO2 [K₃]
- 15. Enlist any two differences between the batch and continuous modes of operation in SuperPro Designer, emphasizing their applications and advantages in material and energy balance analysis. CO3 [K₃]
- 16. How do Gantt charts aid in equipment utilization analysis, illustrating their role in identifying bottlenecks and optimizing process schedules. CO3 [K₂]
- 17. A continuous process is set up for treatment of waste water. Each day, 10⁵ kg cellulose and 10³ kg bacteria enter in the feed stream, while 10⁴ kg cellulose and 1.5x10⁴ kg bacteria leave in the effluent. The rate of cellulose digestion by the bacteria is 7x10⁴ kg/ day. The rate of bacterial growth is 2x10⁴ kg/ day; the rate of cell death by lysis is 5x10² kg/ day. Write balances for cellulose and bacteria in the system. CO4 [K₄]
- 18. How does a lower K_s (*Michaelis-Menten Constant*) value reflect a higher affinity of microorganisms for a substrate? Provide an example or scenario to illustrate the significance of K_s in determining the growth behavior of microorganisms CO4 [K₃]
- 19. Mention any two advantages and limitations of using numerical integration methods such as Euler and Runge Kutta in MATLAB for solving mathematical problems. CO5 [K₂]
- 20. Provide an example of how cost analysis is conducted in bio-chemical engineering flowsheeting, emphasizing the key cost components considered. CO6 [K₃]

PART C (6 x 5 = 30 Marks)

- 21. Assess the importance of incorporating chemical kinetics in bioprocess models. Discuss its implications for process optimization and control. CO1 [K₃]
- 22. How do the fundamental laws, including the continuity equation, energy equation, and equations of motion, are integrated to create comprehensive models for bioprocess systems. Highlight their interdependence and contributions CO2 [K₃]
- 23. In downstream processing of gluconic acid, concentrated fermentation broth containing 20% (w/w) gluconic acid is cooled prior to crystallisation. The concentrated broth leaves an evaporator at a rate of 2000 kg/h and must be cooled from 90°C to 6°C. Cooling is achieved by heat exchange with 2700 kg/h water initially at 2°C. If the final temperature of the cooling water is 50 °C. what is the rate of heat loss from the gluconic acid solution to the surroundings? Assume the heat capacity of gluconic acid is 0.35 cal/g/°C. CO3 [K₄]

Data:

- h (liquid water at 90 °C)= 376.9 kJ/kg
- h (liquid water at 6 °C)= 376.9 kJ/kg
- h (liquid water at 2 °C)= 8.4 kJ/kg
- h (liquid water at 50 °C)= 209.3 kJ/kg

- The rate of heat input is 4.933103 kJ/h
24. Examine the merits and demerits associated with various modes of bioreactors employed in biochemical engineering CO4 [K₂]
25. The synthesis of biomass from glucose (C₆H₁₂O₆) and ammonia (NH₃) is represented by the following generalized equation: CO5 [K₄]



- If the Respiratory Quotient (RQ) for this process is found to be 0.90. Determine the stoichiometric coefficients (a,b,c,d&e) for the given equation.
26. Develop a systematic approach for conducting economic evaluations in biochemical engineering flowsheeting, considering the criteria for assessing project viability. CO6 [K₄]

Answer any FOUR Questions
PART D (4 x 10 = 40 Marks)

27. Tetracycline produced in *Streptomyces aureus* fermentations is purified by crystallisation. One hundred kg of a supersaturated solution containing 7.7 wt% tetracycline is cooled in a batch fluidised-bed crystalliser. Seed crystals of tetracycline are added at a concentration of 40 ppm to promote crystal growth. At the end of the crystallisation process, the remaining solution contains 2.8% tetracycline. CO1 [K₄]

(a) What is the mass of the residual tetracycline solution?

(b) What mass of tetracycline crystals is produced?

28. With a suitable assumption, derive the derivation for Fed batch operation CO2 [K₂]
29. The concentration of salt in a home-made soap maker is given as a function of time by $\frac{dx}{dt} = 3.75 - 3.5x$ CO4 [K₅]

At the initial time, $t = 0$, the salt concentration in the tank is 50 g/L Using Runge-Kutta 4th order method and a step size of, $h = 1.5$ min, what is the salt concentration after 3 minutes?

30. A startup company proposes to use corn stover as a feedstock(39% cellulose, 26 % hemicellulose and 23 % lignin, 12 % ash) to to produce ethanol. The company claims to have discovered a new process that can produce 492 L ethanol/ dry metric ton (130 gal ethanol/dry ton) of feedstock. As a bioprocess engineer, scrutinize the feasibility of the estimates provided by the company . CO5 [K₅]
31. Using PFD and BFD, Illustrate the complete flowsheet for the production of ethanol from lignocellulosic waste . CO6 [K₅]
